## Enantioselective  $\alpha$ -Fluorination and Chlorination of  $\beta$ -Ketoesters by Cobalt Catalyst

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We demonstrated the cobalt-catalyzed asymmetric  $\alpha$ -fluorination and  $\alpha$ -chlorination of  $\beta$ -ketoesters. Both reactions were achieved using a catalytic amount of  $Co(acac)_2$  with  $(R,R)$ -Jacobsen's salen ligand;  $\alpha$ -fluorinated or  $\alpha$ -chlorinated products were thus obtained with a good enentioselectivity.

Chiral fluorinated organic compounds are well recognized as important materials in the field of biological and medicinal chemistry.<sup>1</sup> Recently, the transition metal catalyzed highly enantioselective  $\alpha$ -fluorination of  $\beta$ -ketoesters has been achieved by several groups.2 For example, Togni reported the [TiCl<sub>2</sub>(TADDOLato)]-catalyzed reaction with Selectfluor, and they also discovered a ruthenium catalyst system.<sup>3</sup> Sodeoka demonstrated a Pd/BINAP-catalyzed system with N-fluorobenzenesulfonimide (NFSI).<sup>4</sup> Cahard also described that Cu/Box is an effective catalyst for the  $\alpha$ -fluorination of  $\beta$ -ketoesters.<sup>5</sup> Furthermore, Shibata and Toru attained a high enantioselectivity with a Ni/dbfox catalyst.<sup>6</sup> More recently, a Ni or Mg/N,N,Ntridentate ligand system was reported by Shibatomi and Iwasa,<sup>7</sup> and chiral rare earth perfluorinated organophosphate catalysts were developed by Inanaga.<sup>8</sup> Despite these pioneering studies of enantioselective fluorination, the development of a new catalyst system is still required in this area. Recently, we have been interested in the development of the cobalt-catalyzed asymmetric reaction, and realized the cobalt/pybox-catalyzed asymmetric conjugate addition of thiols to  $\alpha, \beta$ -unsaturated carbonyl compounds.<sup>9</sup> During the course of the cobalt-catalyzed asymmetric reactions, we found that the cobalt/Jacobsen's salen ligand system exhibited a high enantioselectivity for the  $\alpha$ -fluorination of  $\beta$ -ketoesters.

We examined the reaction of ethyl 2-oxocyclopentanecarboxylate (1a) with NFSI using cobalt catalysts.<sup>10</sup> Based on the results of our previous chiral cobalt catalyzed asymmetric reaction,<sup>9</sup> we tested the  $\alpha$ -fluorination reaction of  $\beta$ -ketoesters by  $Co(CIO<sub>4</sub>)<sub>2</sub>•6H<sub>2</sub>O$  with (S,S)-ip-pybox. However, the reaction produced an  $\alpha$ -fluorinated product with a poor result; i.e., a 55% yield and 25% enatiomeric excess (Table 1, Entry 1). Reinvestigation of the effective combination of a cobalt salt and chiral ligand revealed that  $Co(acac)$ , with the  $(R,R)$ -Jacobsen's salen ligand (L2) exhibited a higher enatiomeric excess (60% ee) with almost the same yield (60%) (Entry 4). The enantioselectivity was improved when diethyl ether was used as the solvent, but the yield had decreased to 41% (Entry 5). Fortunately, both the chemical yield and enantioselectivity of the desired products significantly increased at lower reaction temperature (Entries 6 and 7), and the best result was obtained at  $-20\degree\text{C}$  (84% isolated yield with 89% ee). According to the reported results of the metal-catalyzed  $\alpha$ -fluorination of cyclic  $\beta$ -ketoesters by other groups, it seems that moderately bulky groups, such as tertbutyl, at the ester functionality are necessary to attain high Table 1. Cobalt catalysts for the  $\alpha$ -fluorination of ethyl 2oxocyclopentanecarboxylate (1a)<sup>a</sup>





a Reaction conditions: 1a (0.32 mmol), [Co] (0.032 mmol), L1 or L2 (0.032 mmol), NFSI (0.45 mmol), solvent (1.0 mL). <sup>b</sup>Isolated yield. <sup>c</sup>Enantiomeric excess values were determined by GC analysis using Chiraldex G-TA.

enantioselectivity. Actually, most of the reports mainly examined the tert-butyl esters, and there are only two examples of the reaction of the ethyl ester, $3c,8$  which is commercially available. To the best of our knowledge, the highest enantioselectivity reported for the reaction of 1a was 76% ee, and it was attained using a scandium catalyst. It should be emphasized that our cobalt catalyst is superior to the scandium catalyst for the  $\alpha$ fluorination of the ethyl ester **1a** (89% ee, Entry 7).

We used the Co(acac) $\frac{2}{L^2}$  catalyst for the  $\alpha$ -fluorination of other  $\beta$ -ketoesters. These results are summarized in Table 2. The ketoester 1b (methyl ester) produced the desired  $\alpha$ -fluorinated product with 90% ee (Entry 1). The reaction of 1c (tert-butyl ester) also exhibited a good enantioselectivity (86% ee). On the other hand, reduced enantioselectivities were obtained for the reaction of other cyclic  $\beta$ -ketoesters containing six- or sevenmembered rings  $(1d-1f)$  (Entries 3-5). We further examined the reaction of acyclic  $\beta$ -ketoester 1g, but both yield and enantioselectivity were moderate (Entry 6).<sup>11</sup>

Furthermore, the  $Co(acac)<sub>2</sub>/L2$  catalyst worked as a good system for the the enantioselective  $\alpha$ -chlorination of 1a with  $CF<sub>3</sub>SO<sub>2</sub>Cl$  (TFSC) (Scheme 1).<sup>6,12</sup> The reaction was carried out

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	OR	10 mol% Co(acac) <sub>2</sub> 10 mol% L2 OR		
'n		$\gamma_{\rm F}$ NFSI (1.4 equiv) $\omega_{\rm n}$ Et <sub>2</sub> O, 12 h		
<b>1b</b> : $n = 1$ , $R = Me$			$2b-f$	
<b>1c</b> : $n = 1$ , $R = {}^{t}Bu$				
1d: $n = 2$ , $R = Me$		OEt Me		
1e: $n = 2$ , $R = Et$		Me F		
<b>1f</b> : $n = 3$ , $R = Me$		2g		
Me OEt				
Me				
	1g			
Entry	$\mathbf{1}$	Temp/ $\rm ^{\circ}C$	Yield/ $\%$ <sup>b</sup>	$ee/\%$ <sup>c</sup>
1	1 <sub>b</sub>	$-20$	74	90
2	1c	$-20$	65	86
3	1 <sub>d</sub>	0	65	79
$\overline{4}$	1e	0	65	75
5	1f	0	75	79
6	1g	rt	64	71

Table 2. Cobalt-catalyzed  $\alpha$ -fluorination of  $\beta$ -ketoesters (1b- $1g)^a$ 

<sup>a</sup>Reaction conditions:  $\beta$ -ketoester (0.32 mmol), Co(acac)<sub>2</sub> (0.032 mmol), L2 (0.032 mmol), NFSI (0.45 mmol), diethyl ether (1.0 mL). <sup>b</sup>Isolated yield. <sup>c</sup>Enantiomeric excess values were determined by GC analysis with a Chiraldex G-TA for 1b and 1d-1g, or chiral HPLC using Daicel CHIRALPAK AD-H for 1c.





in toluene at room temperature, and the desired chlorinated product was obtained with a 75% ee, but the yield was insufficient (16%). Fortunately, both the yield and enantioselectivity increased to 62% and 88% ee by the addition of molecular sieves 4A.

In conclusion, we demonstrated the cobalt-catalyzed asymmetric  $\alpha$ -fluorination and  $\alpha$ -chlorination of  $\beta$ -ketoesters. Both of these desired reactions were catalyzed by a chiral cobalt catalyst, which was prepared from  $Co(acac)$  with  $(R, R)$ -Jacobsen's salen ligand, and the  $\alpha$ -fluorinated or  $\alpha$ -chlorinated products were obtained with good enantioselectivities.

## References and Notes

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- 10 Typical procedure: A solution of  $Co(acac)_2$  (8.2 mg, 0.032 mmol),  $(R, R)$ -Jacobsen's salen ligand  $(17.5 \text{ mg})$ , 0.032 mmol) and NFSI (141 mg, 0.45 mmol) in anhydrous diethyl ether (1.0 mL) was stirred at  $-20^{\circ}$ C for 10 min. To this solution was added a  $\beta$ -ketoester 1a (50 mg, 0.32 mmol), then stirred for 12 h. Saturated NH4Cl was added for quenching, and the water layer was extracted with diethyl ether  $(1.0 \text{ mL} \times 3)$ . The combined organic layers were washed with brine and dried over MgSO4. Removal of the solvent, followed by flash column chromatography (hexane/ ethyl acetate  $= 2/1$ , afforded the desired product 2a as a colorless oil (47 mg, 84%). The enantiomeric purity was determined to be 89% ee by GC analysis with a Chiraldex G-TA (initial temperature 60 °C, final temperature 165 °C, rate  $3^{\circ}$ C min<sup>-1</sup>, inj. temperature 160 °C, det. temperature 100 °C:  $t(R) = 26.0 \text{ min}, t(S) = 30.5 \text{ min}. [\alpha]_D^{25}$  85.8 (c 0.56, CHCl<sub>3</sub>) (89% ee) {lit.<sup>3c</sup> [ $\alpha$ ]<sup>25</sup> 169.0 (c 1.53, CHCl<sub>3</sub>) (99.7% ee)}.<br><sup>1</sup>H NMP (500 MHz CDCL);  $\delta$  1.32 (t  $I = 7.2$  Hz 3H) <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  1.32 (t,  $J = 7.2$  Hz, 3H), 2.01-2.19 (m, 2H), 2.28-2.39 (m, 1H), 2.48-2.60 (m, 3H), 4.30 (q,  $J = 7.2$  Hz, 2H). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$ 14.00, 18.02 (d,  $J = 2.9$  Hz), 33.88 (d,  $J = 21.1$  Hz), 35.68, 62.33, 94.61 (d,  $J = 199.6$  Hz), 167.44 (d,  $J = 26.8$  Hz), 207.52 (d,  $J = 16.2$  Hz). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>, internal standard:  $C_6F_6$ :  $\delta$  -2.39 (t,  $J = 18.8$  Hz).
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