## Enantioselective $\alpha$ -Fluorination and Chlorination of $\beta$ -Ketoesters by Cobalt Catalyst

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We demonstrated the cobalt-catalyzed asymmetric  $\alpha$ -fluorination and  $\alpha$ -chlorination of  $\beta$ -ketoesters. Both reactions were achieved using a catalytic amount of Co(acac)<sub>2</sub> with (*R*,*R*)-Jacobsen's salen ligand;  $\alpha$ -fluorinated or  $\alpha$ -chlorinated products were thus obtained with a good enentioselectivity.

Chiral fluorinated organic compounds are well recognized as important materials in the field of biological and medicinal chemistry.<sup>1</sup> Recently, the transition metal catalyzed highly enantioselective  $\alpha$ -fluorination of  $\beta$ -ketoesters has been achieved by several groups.<sup>2</sup> For example, Togni reported the [TiCl<sub>2</sub>(TADDOLato)]-catalyzed reaction with Selectfluor, and they also discovered a ruthenium catalyst system.<sup>3</sup> Sodeoka demonstrated a Pd/BINAP-catalyzed system with N-fluorobenzenesulfonimide (NFSI).<sup>4</sup> Cahard also described that Cu/Box is an effective catalyst for the  $\alpha$ -fluorination of  $\beta$ -ketoesters.<sup>5</sup> Furthermore, Shibata and Toru attained a high enantioselectivity with a Ni/dbfox catalyst.<sup>6</sup> More recently, a Ni or Mg/N,N,Ntridentate ligand system was reported by Shibatomi and Iwasa,<sup>7</sup> and chiral rare earth perfluorinated organophosphate catalysts were developed by Inanaga.8 Despite these pioneering studies of enantioselective fluorination, the development of a new catalyst system is still required in this area. Recently, we have been interested in the development of the cobalt-catalyzed asymmetric reaction, and realized the cobalt/pybox-catalyzed asymmetric conjugate addition of thiols to  $\alpha,\beta$ -unsaturated carbonyl compounds.<sup>9</sup> During the course of the cobalt-catalyzed asymmetric reactions, we found that the cobalt/Jacobsen's salen ligand system exhibited a high enantioselectivity for the  $\alpha$ -fluorination of  $\beta$ -ketoesters.

We examined the reaction of ethyl 2-oxocyclopentanecarboxylate (1a) with NFSI using cobalt catalysts.<sup>10</sup> Based on the results of our previous chiral cobalt catalyzed asymmetric reaction,<sup>9</sup> we tested the  $\alpha$ -fluorination reaction of  $\beta$ -ketoesters by  $Co(ClO_4)_2 \cdot 6H_2O$  with (S,S)-ip-pybox. However, the reaction produced an  $\alpha$ -fluorinated product with a poor result; i.e., a 55% yield and 25% enatiomeric excess (Table 1, Entry 1). Reinvestigation of the effective combination of a cobalt salt and chiral ligand revealed that  $Co(acac)_2$  with the (R,R)-Jacobsen's salen ligand (L2) exhibited a higher enatiomeric excess (60% ee) with almost the same yield (60%) (Entry 4). The enantioselectivity was improved when diethyl ether was used as the solvent, but the yield had decreased to 41% (Entry 5). Fortunately, both the chemical yield and enantioselectivity of the desired products significantly increased at lower reaction temperature (Entries 6 and 7), and the best result was obtained at -20 °C (84% isolated yield with 89% ee). According to the reported results of the metal-catalyzed  $\alpha$ -fluorination of cyclic  $\beta$ -ketoesters by other groups, it seems that moderately bulky groups, such as tertbutyl, at the ester functionality are necessary to attain high **Table 1.** Cobalt catalysts for the  $\alpha$ -fluorination of ethyl 2-oxocyclopentanecarboxylate  $(1a)^a$ 



	L1: ( <i>S,S</i> )- <i>ip</i> -pybox	L2: (R,R)-Jacobsen's salen ligand			
Entry	[Co]	L	Solv./ Temp (°C)	Yield /% <sup>b</sup>	ee/% <sup>c</sup>
1	$Co(ClO_4)_2 \cdot 6H_2O$	L1	THF/rt	55	25
2	$Co(ClO_4)_2 \cdot 6H_2O$	L2	THF/rt	50	8
3	$Co(acac)_2$	L1	THF/rt	86	34
4	$Co(acac)_2$	L2	THF/rt	60	60
5	$Co(acac)_2$	L2	Et <sub>2</sub> O/rt	41	73
6	$Co(acac)_2$	L2	$Et_2O/0$	68	85
7	$Co(acac)_2$	L2	$Et_2O/-20$	84	89

<sup>a</sup>Reaction conditions: **1a** (0.32 mmol), [Co] (0.032 mmol), **L1** or **L2** (0.032 mmol), NFSI (0.45 mmol), solvent (1.0 mL). <sup>b</sup>Isolated yield. <sup>c</sup>Enantiomeric excess values were determined by GC analysis using Chiraldex G-TA.

enantioselectivity. Actually, most of the reports mainly examined the *tert*-butyl esters, and there are only two examples of the reaction of the ethyl ester,<sup>3c,8</sup> which is commercially available. To the best of our knowledge, the highest enantioselectivity reported for the reaction of **1a** was 76% ee, and it was attained using a scandium catalyst. It should be emphasized that our cobalt catalyst is superior to the scandium catalyst for the  $\alpha$ fluorination of the ethyl ester **1a** (89% ee, Entry 7).

We used the Co(acac)<sub>2</sub>/L2 catalyst for the  $\alpha$ -fluorination of other  $\beta$ -ketoesters. These results are summarized in Table 2. The ketoester **1b** (methyl ester) produced the desired  $\alpha$ -fluorinated product with 90% ee (Entry 1). The reaction of **1c** (*tert*-butyl ester) also exhibited a good enantioselectivity (86% ee). On the other hand, reduced enantioselectivities were obtained for the reaction of other cyclic  $\beta$ -ketoesters containing six- or sevenmembered rings (**1d–1f**) (Entries 3–5). We further examined the reaction of acyclic  $\beta$ -ketoester **1g**, but both yield and enantioselectivity were moderate (Entry 6).<sup>11</sup>

Furthermore, the Co(acac)<sub>2</sub>/L2 catalyst worked as a good system for the the enantioselective  $\alpha$ -chlorination of 1a with CF<sub>3</sub>SO<sub>2</sub>Cl (TFSC) (Scheme 1).<sup>6,12</sup> The reaction was carried out

1b: n = 1c: n = 1d: n = 1d: n = 1f: n = 0 Me	$O = 1, R = Me$ $i = 1, R = ^{t}Bu$ $i = 2, R = Me$ $i = 2, R = Et$ $i = 3, R = Me$ $O = 0$ $O = 0$ $Me$ $D = 1$	10 mol% Co(ac 10 mol% <b>L2</b> NFSI (1.4 equiv Et <sub>2</sub> O, 12 h	cac) <sub>2</sub> v) Me Me	O F 2b-f O F O Et F 2g
Entry	1	Temp/°C	Yield/% <sup>b</sup>	ee/% <sup>c</sup>
1	1b	-20	74	90
2	1c	-20	65	86
3	1d	0	65	79
4	1e	0	65	75
5	1f	0	75	79
6	1g	rt	64	71

Table 2. Cobalt-catalyzed  $\alpha$ -fluorination of  $\beta$ -ketoesters (1b-1g)<sup>a</sup>

<sup>a</sup>Reaction conditions:  $\beta$ -ketoester (0.32 mmol), Co(acac)<sub>2</sub> (0.032 mmol), L2 (0.032 mmol), NFSI (0.45 mmol), diethyl ether (1.0 mL). <sup>b</sup>Isolated yield. <sup>c</sup>Enantiomeric excess values were determined by GC analysis with a Chiraldex G-TA for 1b and 1d–1g, or chiral HPLC using Daicel CHIRALPAK AD-H for 1c.





in toluene at room temperature, and the desired chlorinated product was obtained with a 75% ee, but the yield was insufficient (16%). Fortunately, both the yield and enantiose-lectivity increased to 62% and 88% ee by the addition of molecular sieves 4A.

In conclusion, we demonstrated the cobalt-catalyzed asymmetric  $\alpha$ -fluorination and  $\alpha$ -chlorination of  $\beta$ -ketoesters. Both of these desired reactions were catalyzed by a chiral cobalt catalyst, which was prepared from Co(acac)<sub>2</sub> with (*R*,*R*)-Jacobsen's salen ligand, and the  $\alpha$ -fluorinated or  $\alpha$ -chlorinated products were obtained with good enantioselectivities.

## **References and Notes**

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- Typical procedure: A solution of Co(acac)<sub>2</sub> (8.2 mg, 10 0.032 mmol), (*R*,*R*)-Jacobsen's salen ligand (17.5 mg, 0.032 mmol) and NFSI (141 mg, 0.45 mmol) in anhydrous diethyl ether (1.0 mL) was stirred at -20 °C for 10 min. To this solution was added a  $\beta$ -ketoester **1a** (50 mg, 0.32 mmol), then stirred for 12h. Saturated NH<sub>4</sub>Cl was added for quenching, and the water layer was extracted with diethyl ether  $(1.0 \text{ mL} \times 3)$ . The combined organic layers were washed with brine and dried over MgSO<sub>4</sub>. Removal of the solvent, followed by flash column chromatography (hexane/ ethyl acetate = 2/1), afforded the desired product **2a** as a colorless oil (47 mg, 84%). The enantiomeric purity was determined to be 89% ee by GC analysis with a Chiraldex G-TA (initial temperature 60 °C, final temperature 165 °C, rate 3 °C min<sup>-1</sup>, inj. temperature 160 °C, det. temperature 100 °C:  $t(R) = 26.0 \text{ min}, t(S) = 30.5 \text{ min}). [\alpha]_D^{25} 85.8 (c \ 0.56, \text{CHCl}_3)$ (89% ee) {lit.<sup>3c</sup>  $[\alpha]_D^{25}$  169.0 (c 1.53, CHCl<sub>3</sub>) (99.7% ee)}. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  1.32 (t, J = 7.2 Hz, 3H), 2.01-2.19 (m, 2H), 2.28-2.39 (m, 1H), 2.48-2.60 (m, 3H), 4.30 (q, J = 7.2 Hz, 2H). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$ 14.00, 18.02 (d, J = 2.9 Hz), 33.88 (d, J = 21.1 Hz), 35.68, 62.33, 94.61 (d, J = 199.6 Hz), 167.44 (d, J = 26.8 Hz), 207.52 (d, J = 16.2 Hz). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>, internal standard: C<sub>6</sub>F<sub>6</sub>):  $\delta$  -2.39 (t, *J* = 18.8 Hz).
- 11 We also examined the fluorination reactions of 2-methoxycarbonyl-1-indanone and 2-ethoxycarbonyl-1-indanone by Co(acac)<sub>2</sub>/L2 catalyst, but the enantioselectivities were 55% ee (98% yield) and 50% ee (94% yield), respectively.
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